

Structure of the Jovian Stratosphere at the Galileo Probe Entry Site

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The structure of the Jovian stratosphere at the Galileo probe entry site is investigated through calculations of radiative heating rates based on measurements of the temperature profile, composition, and aerosol distribution. From analysis of mid-IR observations of Jupiter, we determine a C₂H₂ mole fraction of 1.1–4.3 × 10⁻⁶ at 0.01 mbar, and a C₂H₆ mole fraction of 2.8–6.5 × 10⁻⁶ at 0.4–10 mbar. The derived distributions imply that C₂H₆ and C₂H₂ are the most important coolants in the Jovian stratosphere from 0.004 to 10 mbar, and that the stratosphere is close to radiative equilibrium. In Jupiter's stratosphere, as in the stratosphere of the Earth, photochemical species play an essential role in the energy balance.

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1. INTRODUCTION

Jupiter has a well-developed stratosphere with a mean temperature near 10 mbar, approximately 50 K higher than that at the tropopause. This stratospheric temperature shows remarkably little variation with latitude and longitude. Typical equator-to-pole temperature differences are 2 K or less (Orton *et al.* 1991), although there are some localized auroral phenomena that exhibit larger variations. It is tempting to conclude that energy in the jovian stratosphere is efficiently redistributed by dynamics, perhaps driven by a net radiative imbalance. Yet, it is also possible that momentum sources propagating upward from the troposphere dominate stratospheric dynamics on Jupiter. Despite recent contributions, it is fair to say that the relative roles played by radiation and dynamics, and how these interact with composition and chemistry to determine stratospheric structure, are not well determined. A critical factor in all stratospheric investigations is local radiative balance. It is this subject that concerns us here.

The temperature profile in the jovian stratosphere has been studied through observations of emissions from the ν_4 band of

CH₄ (Gillet *et al.* 1969, Orton *et al.* 1991), stellar occultation experiments (Ververka *et al.* 1974, Hubbard *et al.* 1972, 1995), and spacecraft radio occultations experiments (Lindal *et al.* 1981). More recently, the Atmospheric Structure Instrument (ASI) on the Galileo probe (Sieff *et al.* 1997, 1998) determined the temperature profile in the jovian stratosphere through analysis of the probe deceleration as it passed through the atmosphere. These *in situ* results have a higher fidelity than previous determinations of the temperature profile because they are free of the assumptions that accompany interpretation of any remote sensing measurement. In addition, numerous UV and IR observations of Jupiter have been conducted in recent years, in connection with the SL9 impacts and in support of the Galileo mission. These observations constrain the gaseous and aerosol abundances in the jovian stratosphere. Thus, conditions are propitious for a new and in-depth investigation of stratospheric radiation balance on Jupiter.

We examine the energy balance of the stratosphere at the probe entry site by calculating heating rates using the latest CH₄ absorption coefficients and cooling rates and considering realistic altitude distributions of C₂H₂ and C₂H₆ to calculate heating rates. To determine the altitude distributions of C₂H₂ and C₂H₆, we use mid-IR spectra of equatorial regions in addition to results from previous analyses of mid-IR and UV observations. Comparison of the net heating and cooling rates demonstrates that the jovian stratosphere is close to radiative equilibrium. Or, more precisely, the departures from radiative equilibrium appear to be smaller than the uncertainty in the calculated rates. We find that C₂H₂ and C₂H₆ are the primary coolants in the stratosphere with C₂H₆ dominating at most pressures. The thermal balance and photochemistry of the stratosphere are closely coupled. Latitudinal gradients in the C₂H₂ and C₂H₆ densities may cause radiative forcing of wind systems that will in turn redistribute C₂H₂ and C₂H₆. Knowledge of composition, aerosol distribution, and temperature are necessary to develop a complete understanding of the jovian stratosphere.

2. PREVIOUS RADIATIVE EQUILIBRIUM MODELS

Cess and Khetan (1973) and Wallace *et al.* (1974) conducted the first successful radiative equilibrium calculations of the thermal profile in the jovian stratosphere. The thorough study by Wallace *et al.* (1974) succeeds in identifying the near-IR CH₄ bands as the main source of heating in the stratosphere. Models including these bands alone and cooling through the ν_4 band of CH₄ matched the measurements of the stratospheric temperature available at that time. Wallace *et al.* (1974) did not include radiative cooling by C₂H₂ and C₂H₆ in their calculations but estimated that significant amounts of these molecules could cool the upper stratosphere by tens of Kelvins.

Cess and Chen (1975) improved the models of Cess and Khetan (1973) to include cooling by C₂H₂ and C₂H₆. Using constant mole fractions of 1×10^{-5} and 5×10^{-7} respectively, Cess and Chen (1975) found that C₂H₂ and C₂H₆ cooled the stratosphere by 20 K. This caused the models to be cooler than temperatures inferred from observations. In order to bring the models back into agreement with the observations, Cess and Chen (1975) hypothesized the presence of significant aerosol heating. Stratospheric aerosols had previously been postulated to explain the low UV albedo of Jupiter (Axel 1972). However, the aerosol heating in these calculations was not deduced from observations but was parameterized and adjusted to produce agreement between observed and calculated temperature profiles.

Appleby and Hogan (1984) and Appleby (1990) continued studies of Jupiter's thermal structure, motivated partly by the availability of Voyager measurements. Following Cess and Chen (1975), Appleby and Hogan (1984) included aerosol heating in a parameterized fashion and found that the aerosols must absorb 3.8% of the total solar flux in order to bring the models into agreement with temperatures inferred from the Voyager radio occultation experiment. The difference between Appleby and Hogan's aerosol model and their dust-free model is 10 K. C₂H₂ and C₂H₆ were included in these models using the formalism developed by Cess and Chen (1975). Constant mixing ratios were assumed with values of 10^{-5} and 5×10^{-7} for C₂H₆ and C₂H₂. The relative importance of cooling by CH₄, C₂H₂ and C₂H₆ was not discussed.

More recent work on the thermal structure of the jovian stratosphere has emphasized the role of dynamics. Voyager and ground-based observations indicate that the temperature of the stratosphere is surprisingly uniform with latitude, implying efficient meridional redistribution of heat (Conrath *et al.* 1990, Orton *et al.* 1991). Conrath *et al.* (1990) utilized a model including radiation and friction to study the dynamics of outer planet stratospheres. These authors found that the lower stratosphere is forced primarily by the prevailing winds in the upper troposphere. Radiative forcing becomes important in the upper stratosphere causing a Hadley-cell circulation pattern with upwelling at low latitudes and subsidence at high latitudes, driven by the excess solar energy deposited in equa-

torial regions. Conrath *et al.* (1990) did not consider aerosol heating.

West *et al.* (1992) studied the radiative and dynamical structure of the jovian stratosphere using a Transformed Eulerian Mean model to infer diabatic circulation patterns and eddy forcing from observed temperatures and calculations of net radiative heating rates. These authors utilized improved band parameters to calculate energy deposition in CH₄ and paid particular attention to aerosols. Observations from IUE and Voyager were used to determine the aerosol distribution as a function of latitude. Cooling from C₂H₂ and C₂H₆ emissions was included but the C₂H₂ and C₂H₆ distributions were assumed to be constant with latitude and altitude. West *et al.* (1992) found that the preponderance of aerosols at high latitudes forced a circulation pattern opposite to that calculated by Conrath *et al.* (1990) with subsidence in equatorial regions. The results from West *et al.* (1992) were used by Friedson *et al.* (1999) to show that the meridional mixing rate of aerosols created in the SL9 collisions was consistent with horizontal eddy mixing derived from the divergence of the Eliassen–Palm flux.

Moreno and Sedano (1997) performed dynamical calculations similar to those of West *et al.* (1992) but utilized an aerosol distribution based on analysis of multi-wavelength HST images of Jupiter. These data provide higher spatial resolution than possible with the IUE and Voyager data and the aerosol distribution derived by Moreno and Sedano (1997) has fewer aerosols and therefore less heating than that of West *et al.* (1992) at high northern latitudes. The treatment of gaseous opacities was similar in Moreno and Sedano (1997) and West *et al.* (1992), with the exception that the former authors included radiative cooling by NH₃, which was neglected in earlier studies. Moreno and Sedano (1997) followed earlier authors in assuming the distributions of C₂H₂ and C₂H₆ to be uniform with latitude and altitude. Moreno and Sedano (1997) found good agreement with West *et al.* (1992) at pressures greater than 100 mbar but a significantly different circulation pattern at lower pressures, as a consequence of the different aerosol heating pattern.

The dynamical studies mentioned above have demonstrated the important role of latitudinal variations in aerosol heating in driving dynamics. However, all studies to date have made the questionable assumption that the mole fractions of C₂H₂ and C₂H₆ are constant with latitude and most have incorrectly assumed that they are constant with altitude. Our results show that latitudinal variations in C₂H₆ could be more important than aerosol variations in driving stratospheric dynamics.

3. OBSERVATIONAL CONSTRAINTS

The radiative terms in the energy balance of Jupiter's stratosphere depend on the abundance of gaseous constituents, the abundance, size distribution, and optical properties of aerosols, and the temperature profile. We discuss all of these quantities, starting with the temperature profile.

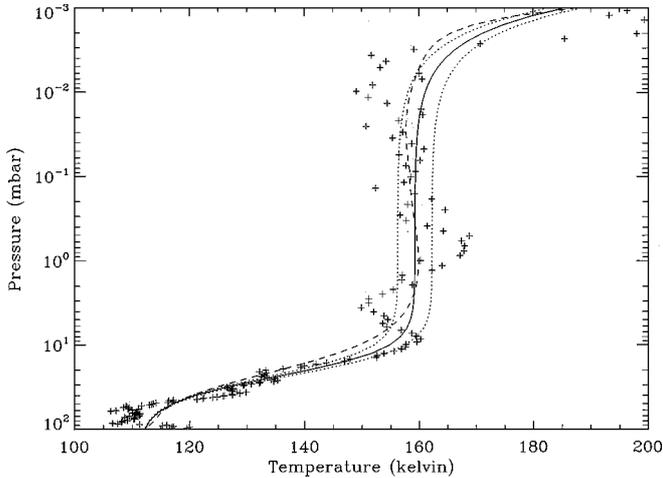


FIG. 1. The temperature profile inferred from the Galileo ASI experiment and three analytic approximations to the mean thermal structure. The solid line is the best-fit least-squares model given in Eq. (1). The dotted lines are models with T_1 altered by ± 3 K from the best fit value of 160 K. The dashed line is a radiative equilibrium model.

3.1. Temperature Profile

Figure 1 shows the temperature profile of the Jovian stratosphere derived from ASI measurements of the acceleration the Galileo probe (Sieff *et al.* 1998). These data were obtained at a latitude of 6°N . The mean temperature in the stratosphere is 160 K and the temperature at the tropopause is 110 K. The temperature rise to the thermosphere begins at about 0.001 mbar. Oscillations in the measured temperature about the mean value are significantly larger than the uncertainty in the ASI measurements. The temperature perturbations have been interpreted as the manifestation of waves in the atmosphere (Sieff *et al.* 1997, 1998, Young *et al.* 1997). We are interested in the mean state of the atmosphere and have therefore fit the ASI temperature profile with a smooth function of the form

$$T(p) = T_0 + \frac{T_1 - T_0}{1 + (p/p_1)^{\alpha_1}} + \frac{T_2 - T_1}{1 + (p/p_2)^{\alpha_2}}. \quad (1)$$

The constants determined by the fit have the following values: $T_0 = 111$ K, $T_1 = 159.7$ K, $p_1 = 24.1$ mbar, $\alpha_1 = 2.45$, $T_2 = 900$ K, $p_2 = 6.13 \times 10^{-5}$ mbar, and $\alpha_2 = 1.21$. We use the fitted profile in the analysis of observations of C_2H_2 and C_2H_6 emissions and in the calculation of radiative heating and cooling rates to be discussed below. Also shown in Fig. 1 are analytic models with T_1 differing by 3 K from the best fit value. These curves are used to evaluate the uncertainty in the mole fractions and cooling rates.

Horizontal variations in the stratospheric temperature have been addressed by Orton *et al.* (1991), who published maps of brightness temperature in the ν_4 band of CH_4 . These observations revealed that the zonal mean structure alternates between states of warm subtropical ($\sim 14^\circ$) and cool equatorial regions

to cool subtropical and warm equatorial regions, with a period of about four years. The brightness temperature variations from equator to $\sim 14^\circ$ have a magnitude of 1–2 K. Leovy *et al.* (1991) and Friedson (1999) have interpreted these variations as an oscillation in the jovian stratosphere similar to the quasi-biennial oscillation (QBO) in the terrestrial stratosphere. Friedson (1999) also points out that the variations in the physical temperature of the stratosphere are larger than the variation in brightness temperature observed by Orton *et al.* (1991) because the observations sample a broad pressure region and the temperature resulting from the QBO-like oscillation is confined to a relatively narrow pressure region. If Friedson's interpretation is correct then some of the apparent perturbations in the ASI profile are long-lived and should be considered in our analysis. However, Friedson calculates temperature variations from 7.2°N to 14°S of about 3 K and the possibility of that the wave-like features are long-lived is addressed by considering the range of temperature profiles discussed above.

3.2. The CH_4 and He Distributions

Of the numerous species detected in the atmosphere of Jupiter, H_2 , CH_4 , C_2H_2 , and C_2H_6 are important to the energy balance in the stratosphere. Measurements of the abundances of the thermally active species in the equatorial stratosphere are summarized in Table I.

The Gas Chromatograph Mass Spectrometer Experiment (GCMS) on the Galileo probe measured the He and CH_4 abundances in the jovian troposphere. Helium is relatively abundant with a mole fraction of 13.6%, but is inert and influences the atmosphere primarily by altering the mean molecular weight. A CH_4 mole fraction of 0.2% was measured in the troposphere. Both He and CH_4 should have constant mole fractions at pressures greater than ~ 0.01 mbar. At lower pressures their abundance decreases because of diffusive separation. The CH_4 profile near the homopause has been characterized by Yelle *et al.* (1996) through analysis of the Voyager 2 Ultraviolet Spectrometer (UVS) occultation of α Leo.

In the appendix we derive the following formula for the diffusive equilibrium distribution of a minor species,

$$f_i(p) = f_{i0}(1 + r(p_0/p)^{1-\gamma})^{\frac{1-m_i/m_0}{1-\gamma}}, \quad (2)$$

where $f_i(p)$ is the mole fraction of the i th minor constituent and $f_{i0}(p)$ is its value deep in the atmosphere. The quantity r is the ratio of the eddy diffusion and molecular diffusion coefficients at reference pressure p_0 ; m_i is the molecular mass of CH_4 ; and m_0 the molecular mass of the background atmosphere. The eddy diffusion coefficient is assumed to have the same temperature dependence as the molecular diffusion coefficient and to vary with pressure as

$$K(p) = K_0(p_0/p)^\gamma. \quad (3)$$

Equation (2) differs from previous published solutions because

TABLE I
Measurements of Stratospheric Composition on Jupiter

Species	Mole Fraction	Pressure ^a (mbar)	Technique/ Instrument ^b	Assumed Temperature	Reference
H ₂	0.87	>100	GCMS	—	Niemann <i>et al.</i> (1996)
He	0.13	>100	GCMS	—	Niemann <i>et al.</i> (1996)
CH ₄	1–2 × 10 ⁻⁴	1–4 × 10 ⁻⁴	UVS	—	Yelle <i>et al.</i> (1996)
	1.9 × 10 ⁻³	>100	GCMS	—	Niemann <i>et al.</i> (1996)
C ₂ H ₂	1 × 10 ⁻⁷		IUE	—	Clarke <i>et al.</i> (1982)
	9–11 × 10 ⁻⁸	5–15 (10)	IUE	—	Gladstone and Yung (1983)
	2–4 × 10 ⁻⁸	5–15 (10)	IUE	—	Wagener <i>et al.</i> (1985)
	7–13 × 10 ⁻⁸	0.1–4 (1.5)	IR/FP	150–170 K	Noll <i>et al.</i> (1986)
	3.6 × 10 ⁻⁸	5–15 (10)	IUE	—	McGrath <i>et al.</i> (1990)
	3.6–4.3 × 10 ⁻⁸	5–15 (10)	HUT(1)	—	Morrissey <i>et al.</i> (1995)
	2.6–3.1 × 10 ⁻⁸	5–15 (10)	HUT(2)	—	
	1.8–2.8 × 10 ⁻⁸	8	CELESTE	140 K	Sada <i>et al.</i> (1998)
	1–3 × 10 ⁻⁸	20–60	FOS	—	B�etremieux and Yelle (2000)
	2–5 × 10 ⁻⁸	10	FOS	—	
	8.3–10 × 10 ⁻⁷	0.3	ISO	165 K	Fouchet <i>et al.</i> (2000)
	1.0–1.3 × 10 ⁻⁷	4	ISO	165 K	Fouchet <i>et al.</i> (2000)
	1.1–4.3 × 10 ⁻⁶	0.01 ^c	Irshell	160 K	This work
C ₂ H ₆	4–7 × 10 ⁻⁶	0.1–4 (1.5)	IR/FP	150–170 K	Noll <i>et al.</i> (1986)
	2.2–3.4 × 10 ⁻⁶	0.7–30 (5)	IR/Heterodyne	150–170 K	Kostiuk <i>et al.</i> (1987)
	1–12 × 10 ⁻⁶	1–20 (4)	IUE	—	Gladstone and Yung (1983)
	1–4 × 10 ⁻⁶	1–20 (4)	IUE	—	Wagener <i>et al.</i> (1985)
	1.7 × 10 ⁻⁶	1–20 (4)	IUE	—	McGrath <i>et al.</i> (1990)
	2.1–3.9 × 10 ⁻⁶	1–20 (4)	HUT(1)	—	Morrissey <i>et al.</i> (1995)
	2.4–3.8 × 10 ⁻⁶	1–20 (4)	HUT(2)	—	Morrissey <i>et al.</i> (1995)
	2.6–5.8 × 10 ⁻⁶	5	CELESTE	145 K	Sada <i>et al.</i> (1998)
	0.8–1.2 × 10 ⁻⁵	1	ISO	165 K	Fouchet <i>et al.</i> (2000)
	2.0–3.1 × 10 ⁻⁶	10	ISO	152 K	Fouchet <i>et al.</i> (2000)
	2.8–6.5 × 10 ⁻⁶	0.4–10 (5)	Irshell	160 K	This work

^a The range of pressures indicates the width of the contribution function; the value in parentheses, the location of the peak. Mole fractions are reported at the peak contribution function pressure. The mole fractions of H₂, He, and CH₄ are expected to be constant at pressures greater than ~10 μbar.

^b GCMS, Galileo Probe Gas Chromatograph and Mass Spectrometer experiment; UVS, Voyager Ultraviolet Spectrometer occultation experiment; IUE, The International Ultraviolet Explorer; IR/FP Ground-based IR observations with a Fabry–Perot Spectrometer with resolving power of 7100; HUT, the Hopkins Ultraviolet Telescope—(1) and (2) refer to the first and second flight; FOS, the Faint Object Spectrograph on HST; ISO, the Infrared Spectroscopy Observer; Irshell, The University of Texas mid-IR spectrometer; IR/Heterodyne, IR observations with a Heterodyne Spectrometer at a resolving power of 10⁶.

^c Derivation of this value assumes C₂H₂ mole fractions of 8.3–10 × 10⁻⁷ and 2–4 × 10⁻⁸ at 0.3 and 10 mbar.

it has no explicit dependence on temperature; as a consequence, it is particularly useful for thermal balance calculations. This formula neglects photolysis, but this is not a major concern because the diffusion time constant is more than a factor of two less than the photolysis time constant at all levels. Moreover, the uncertainty in the rate of diffusive separation is larger than the effect of photolysis, so nothing is gained by including chemistry until the altitude profile of CH₄ is better constrained.

Figure 2 shows a CH₄ distribution based on Eq. (2), the Voyager UVS results and the GCMS measurement along with a CH₄ profile presented in Gladstone *et al.* (1996) for the north equatorial belt. The Gladstone *et al.* (1996) model has been translated to the smoothed ASI temperature profile by assuming that mole fractions at a given pressure remain constant despite the different temperature profiles used. The onset of diffusive equilibrium occurs at a deeper pressure in the Gladstone *et al.* (1996) model than indicated by the UVS determination of the CH₄ density (Yelle *et al.* 1996). Thus, although the Gladstone *et al.* (1996) calculations are far more sophisticated, we adopt the

analytic models for our thermal structure studies. The diffusive equilibrium model is also used to model the He mole fraction. Molecular diffusion coefficients for CH₄ and He are taken from Mason and Marrero (1970).

3.3. Previous Determinations of C₂H₂ and C₂H₆ Mole Fractions

UV observations have been used to infer the distributions of C₂H₂ and C₂H₆ in the jovian stratosphere for nearly 20 years. Gladstone and Yung (1983) analyzed the International Ultraviolet Explorer (IUE) observations of Jupiter from Clarke *et al.* (1982) at wavelengths less than 174 nm and, assuming a homogeneous atmosphere, determined C₂H₂ and C₂H₆ mole fractions of 10⁻⁷ and 7 × 10⁻⁶, respectively. Wagener *et al.* (1985) also analyzed IUE observations of Jupiter but extended the analysis to the 145–315 nm region and determined mole fractions for C₂H₂ of 3 ± 1 × 10⁻⁸, assumed to be constant in the stratosphere. Wagener *et al.* (1985) argue that the difference between

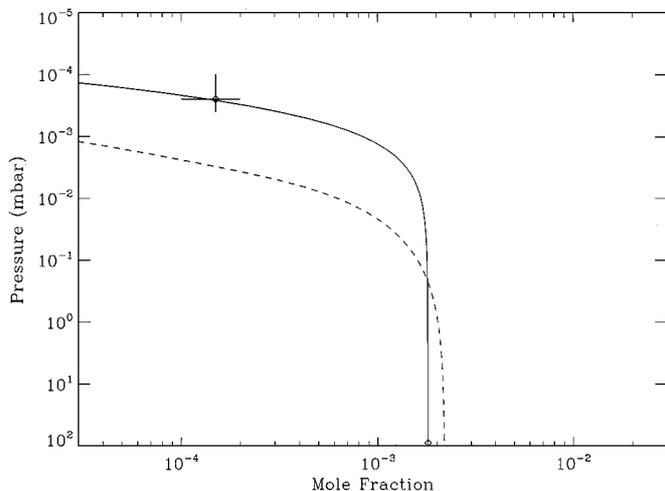


FIG. 2. The distribution of CH_4 . The solid line is constrained to fit the inferred mole fractions of Niemann *et al.* (1996) and Yelle *et al.* (1996) from Table I and is the model used in this investigation. The dashed line is the NEB model from Gladstone *et al.* (1996) and provides a poor fit to the Yelle *et al.* (1996) results.

their results and Gladstone and Yung (1983) is due to the restricted wavelength range in the latter study. This is probably correct, but it means that the Gladstone and Yung (1983) results pertain to higher altitudes than the Wagener *et al.* (1985) results, not that the Gladstone and Yung (1983) results should be supplanted by the Wagener *et al.* (1985) results, as those authors suggest. Taken together the results from Gladstone and Yung (1983) and Wagener *et al.* (1985) indicate that the C_2H_2 mole fraction increases strongly with decreasing pressure in the jovian atmosphere.

Gladstone and Yung (1983) also provided a useful analytic approximation for the relationship between reflectivity and single

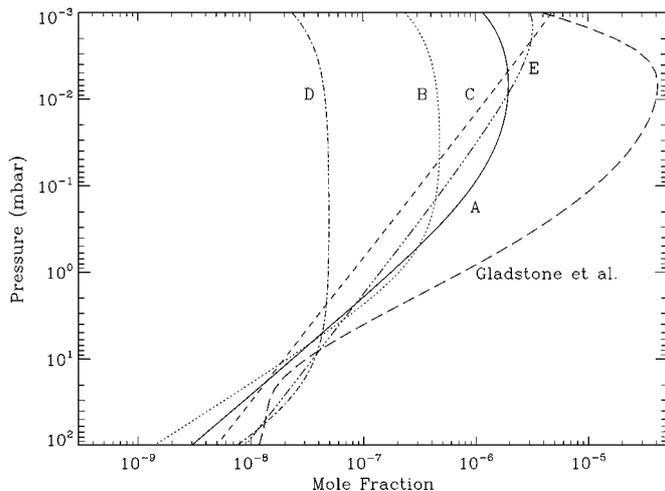


FIG. 3. Models for the C_2H_2 distribution that fit the Irshell observations of the ν_5 band. Parameters describing the models are summarized in Table II. The NEB model from Gladstone *et al.* (1996) is also shown.

scattering albedo in a homogeneous atmosphere from which the molecular abundances can be obtained. McGrath *et al.* (1990) in an analysis of later IUE data and Morrissey *et al.* (1995) in an analysis of data from the Hopkins Ultraviolet Telescope (HUT) used this formula. The former authors determined C_2H_2 and C_2H_6 mole fractions of 3.6×10^{-8} and 1.7×10^{-6} , while the latter authors determined C_2H_2 and C_2H_6 mole fractions of $3.6\text{--}4.3 \times 10^{-8}$ and $2.1\text{--}3.9 \times 10^{-6}$ for data acquired during the Astro-1 mission and $2.6\text{--}3.1 \times 10^{-8}$ and $2.4\text{--}3.8 \times 10^{-6}$ for data acquired during the Astro-2 mission. The HUT observations have good signal-to-noise from 145–190 nm making these data particularly useful for retrieval of the C_2H_6 abundance. Morrissey *et al.* (1995) do not discuss the pressure regime investigated by their observations, but because their wavelength range is similar to that of Gladstone and Yung (1983), it is likely that the same pressures are probed.

Bétreminx and Yelle (1999, 2000) analyzed observations of Jupiter in the 175–230 nm spectral range made with the Faint Object Spectrograph (FOS) on the Hubble Space Telescope (HST). Their radiative transfer models included Raman scattering and utilized vertically inhomogeneous aerosol layers, clouds, and distributions of absorptive gaseous constituents. Bétreminx and Yelle (1999, 2000) found a cloud structure significantly different from that found by Wagener *et al.* (1985); however, the C_2H_2 mole fraction in the 20–60 mbar region was determined to be $1\text{--}3 \times 10^{-8}$, consistent with Wagener *et al.*'s (1985) results. Bétreminx and Yelle (1999, 2000) found that the C_2H_2 mole fraction increased to $2\text{--}5 \times 10^{-8}$ near 10 mbar indicating an increase of mole fraction with decreasing pressure, as suggested above. The relatively high low wavelength limit of valid FOS data prevented probing any higher into the stratosphere or determining the C_2H_6 mole fraction. Edgington *et al.* (1998) also analyzed the HST/FOS spectra of Jupiter; however, the C_2H_2 distribution is not discussed in depth in this paper and the fit to

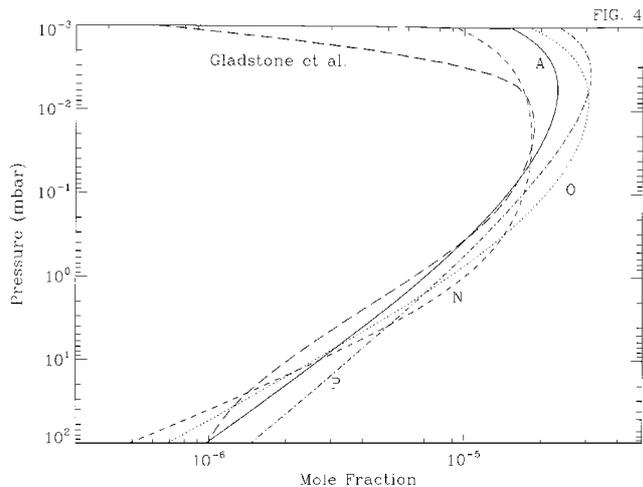


FIG. 4. Models for the C_2H_6 distribution that fit the Irshell observations of the ν_9 band. Parameters describing the models are summarized in Table II. The NEB model from Gladstone *et al.* (1996) is also shown.

the C_2H_2 band depths is poor in the 175–200 nm region (see their Figs. 4 and 5).

The first observation of C_2H_2 and C_2H_6 mid-IR emissions from Jupiter is due to Ridgway (1974), followed quickly by Combes *et al.* (1974). Subsequent observations were published by Orton and Aumman (1977), Tokunaga *et al.* (1976, 1979, 1979), Encrenaz *et al.* (1978), Noll *et al.* (1986), and Kostiuk *et al.* (1987). More recently, Sada *et al.* (1998) analyzed mid-IR observations of Jupiter obtained in late 1994 and early 1995, shortly after the collision of comet P/Shoemaker–Levy 9. Observations were made with the Celeste spectrometer with the slit aligned along the central meridian. The data analyzed extended from $58^\circ S$ to $58^\circ N$. Sada *et al.* (1998) determined stratospheric temperatures through analysis of CH_4 emissions and derived values ~ 10 K warmer than those measured by Galileo. The authors suggest that the discrepancy is evidence for horizontal variations in the stratospheric temperature, but this suggestion is inconsistent with the magnitude of stratospheric temperature variations observed by Orton *et al.* (1991). Sada *et al.* (1998) determined C_2H_2 and C_2H_6 densities by using altitude distributions calculated with photochemical models that were scaled to produce agreement with the observations. They determine mole fractions for C_2H_2 and C_2H_6 of $2.3 \pm 0.5 \times 10^{-8}$ and $3.9_{-1.3}^{+1.9} \times 10^{-6}$ at 8 and 5 mbar, respectively. These values should be viewed as an average over equatorial and midlatitude regions. If the stratospheric temperature is indeed cooler than found by Sada *et al.* (1998) then the mole fractions of C_2H_2 and C_2H_6 would be larger.

Fouchet *et al.* (2000) derived the abundance of C_2H_2 and C_2H_6 from analysis of ISO observations of Jupiter. These observations encompassed a region on Jupiter extending from $30^\circ S$ to $30^\circ N$ and $\pm 20^\circ$ in longitude about the central meridian. These authors determined the stratospheric temperature from observations of the CH_4 emissions and found a profile that was 2 K cooler than the smoothed ASI profile from 5 to 30 mbar and 4 K warmer at pressures less than 5 mbar. Fouchet *et al.* (2000) considered C_2H_2 and C_2H_6 profiles in which the mole fraction varies as pressure to a constant power. They determined a C_2H_2 mole fraction that varies from $8.9_{-0.6}^{+1.1} \times 10^{-7}$ at 0.3 mbar to $1.1_{-0.1}^{+0.2} \times 10^{-7}$ at 4 mbar and a C_2H_6 mole fraction that varies from $1.0 \pm 0.2 \times 10^{-5}$ at 1 mbar to $2.6_{-0.6}^{+0.5} \times 10^{-6}$ at 10 mbar. Fouchet *et al.* (2000) determine an uncertainty in the mole fractions of C_2H_2 and C_2H_6 of 20 and 25% for a temperature uncertainty of 2 K.

The preceding summary reveals that, although there is agreement among most of the determinations of C_2H_2 and C_2H_6 mole fractions to within a factor of several, it is difficult to use these observations to construct precise altitude distribution models. Several of the observations encompass a large range of latitudes, while it is our intention to concentrate on the equatorial region. In addition, different studies assume different stratospheric temperature profiles and, for the mid-IR measurements, this has a large effect on the derived mole fractions. Similarly, different investigators characterize the altitude distribution of species in different ways, with some adopting constant profiles, others using complex photochemical models, and others (including us)

using simple analytic forms. Making sense of these different results is made more difficult by the fact that the spectra are formed over a broad altitude region. Often, it is not clear if discrepant results represent a real difference or a different altitude weighting. For our purposes, it is essential that consistent assumptions are made in the analysis of the hydrocarbon emissions and the radiative balance calculations, and that the uncertainties in these quantities are understood. Therefore, we present a new determination of the C_2H_2 and C_2H_6 density profiles through analysis of emissions from the equatorial regions obtained using a small aperture and high spectral resolution. The results, presented in Figs. 3 and 4 are described in the next section.

3.4. Irshell Measurements of C_2H_2 and C_2H_6

Spectra of the the ν_5 band of C_2H_2 (Fig. 5) and the ν_9 band of C_2H_6 (Fig. 6) were recorded at NASA's Infrared Telescope Facility with the Irshell spectrometer (Lacy *et al.* 1989) in 1995, on May 16 (UT 12:02) and May 18 (UT 13:24), respectively. These data, selected from observations published by Bézard *et al.* (1997a), sample a $3 \times 3''$ region centered at $2^\circ N$ latitude, for C_2H_2 and $21^\circ S$ latitude, for C_2H_6 , corresponding to jovian air masses of 1.01 and 1.06. The spectra have resolutions of 0.11 (C_2H_2) and 0.16 cm^{-1} (C_2H_6).

The observations are analyzed assuming the temperature profiles discussed in Section 3.1, considering both the best-fit temperature profile and the profiles that differ by ± 3 K. We examine this range of thermal profiles to take into account both the observed temperature variations with latitude and possible variations in the vertical structure of the thermal profile. An alternate approach would be to construct thermal profiles based on Friedson's (1999) theory of the stratospheric thermal profile. We choose not to follow this course for several reasons. The arguments of Leovy *et al.* (1991) and Friedson (1999), though

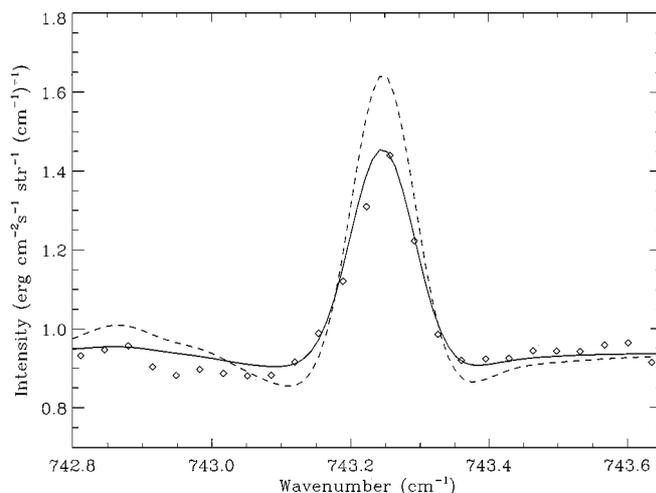


FIG. 5. Irshell observations of C_2H_2 and a synthetic spectrum (solid line) based on the smoothed ASI temperature profile and model A. The dashed line is a synthetic spectrum based on the Gladstone *et al.* (1996) photochemical model.

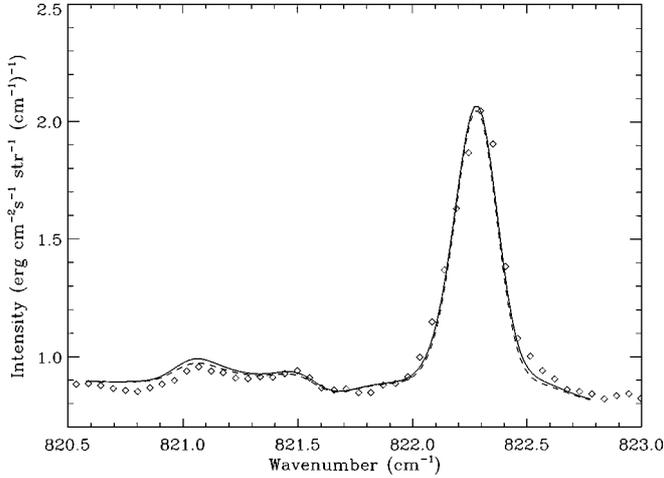


FIG. 6. Irshell observations of C_2H_6 and a synthetic spectrum (solid line) based on the smoothed ASI temperature profile and model A. The dashed line is a synthetic spectrum based on the Gladstone *et al.* (1996) photochemical model, modified at low pressure to be consistent with the eddy diffusion coefficient from Yelle *et al.* (1996).

compelling, are not conclusive. For example, they are unable to tie the QBO-like oscillation directly to a forcing mechanism, as has been done for the terrestrial stratosphere (Friedson 1999). Therefore, we believe that it is wiser to treat the QBO explanation as a hypothesis, rather than incorporate it directly into our work. Moreover, Friedson's calculations indicate that the largest temperature differences between $6^\circ N$ latitude (the region sampled by ASI measurements) and the $2^\circ N$ and $21^\circ S$ latitude (where the C_2H_2 and C_2H_6 observations probed) are 7 and 3 K, respec-

tively. While the 7 K difference lies outside our error margin, it also pertains to a narrow pressure region. The more significant coolant, C_2H_6 , is analyzed with a thermal profile within the theoretically predicted variation at all pressures. The effects of these departures from our assumed smooth profile would not produce distributions of C_2H_2 , and particularly C_2H_6 , very different from those derived here.

In order to model the vertical distributions of C_2H_2 and C_2H_6 , we use a function of the form

$$f = \frac{f_1 f_2}{f_1 + f_2}, \quad (4)$$

where f_1 has the form given in Eq. (2) with the same exponential factors, but f_{10} is allowed to differ from that for CH_4 . The function f_2 is given by

$$f_2(p) = f_{20}(p_0/p)^\gamma. \quad (5)$$

Thus, the function $f(p)$ has three free parameters, f_{10} , f_{20} , and γ that control, respectively, the mole fraction in the upper stratosphere, the mole fraction at the base of the stratosphere, and the slope in the intervening region. This provides enough freedom to match the C_2H_2 and C_2H_6 observations and to mimic the more complete photochemical calculations. Figures 3 and 4 show a variety of models that fit the Irshell observations. Values for the parameters needed to calculate these distributions are presented in Table II.

Synthetic spectra are computed with radiative transfer calculations that use a line-by-line (LBL) technique to determine absorption coefficients based on line positions, energy levels,

TABLE II
Composition Models

Model	C_2H_2			C_2H_6			Temp. Profile
	f_{10}	f_{20}	γ	f_{10}	f_{20}	γ	
A	2.5×10^{-6}	3.0×10^{-9}	0.9	2.0×10^{-5}	4.0×10^{-7}	0.80	med
B	5.0×10^{-7}	1.4×10^{-9}	1.2	2.0×10^{-5}	4.0×10^{-7}	0.80	med
C	2.0×10^{-4}	3.0×10^{-9}	0.6	2.0×10^{-5}	4.0×10^{-7}	0.80	med
D	5.0×10^{-8}	9.0×10^{-9}	1.2	2.0×10^{-5}	4.0×10^{-7}	0.80	med
E	1.0×10^{-5}	9.0×10^{-9}	0.6	2.0×10^{-5}	4.0×10^{-7}	0.80	med
F	5.0×10^{-6}	1.4×10^{-9}	1.2	4.0×10^{-5}	1.2×10^{-6}	0.60	cold
G	4.0×10^{-4}	5.0×10^{-9}	0.6	4.0×10^{-5}	1.2×10^{-7}	0.60	cold
H	2.0×10^{-7}	3.0×10^{-8}	1.2	4.0×10^{-5}	1.2×10^{-7}	0.60	cold
I	2.0×10^{-5}	1.0×10^{-8}	0.6	4.0×10^{-5}	1.2×10^{-7}	0.60	cold
J	1.0×10^{-7}	1.4×10^{-9}	1.2	1.6×10^{-5}	3.0×10^{-7}	0.80	hot
K	3.0×10^{-5}	5.0×10^{-9}	0.6	1.6×10^{-5}	3.0×10^{-7}	0.80	hot
L	3.5×10^{-8}	9.0×10^{-9}	1.2	1.6×10^{-5}	3.0×10^{-7}	0.80	hot
M	3.0×10^{-7}	8.0×10^{-9}	0.6	1.6×10^{-5}	3.0×10^{-7}	0.80	hot
N	2.5×10^{-6}	3.0×10^{-9}	0.9	3.5×10^{-5}	1.0×10^{-6}	0.47	med
O	2.5×10^{-6}	3.0×10^{-9}	0.9	4.0×10^{-5}	7.0×10^{-7}	0.60	med
P	2.5×10^{-6}	1.4×10^{-9}	0.9	6.0×10^{-5}	1.5×10^{-6}	0.40	med
Q	3.5×10^{-6}	4.5×10^{-9}	0.9	4.0×10^{-5}	1.2×10^{-6}	0.60	cold
R	3.5×10^{-6}	2.0×10^{-9}	0.9	1.6×10^{-5}	3.0×10^{-7}	0.80	hot

Note: All models use $p_0 = 0.001$ mbar and $K(p_0) = 7 \times 10^5$ $cm^2 s^{-1}$.

and strengths from the GEISA database (Husson *et al.* 1991). The line profile is represented with a Voigt function. We use a collisional broadening coefficient of 0.075 and 0.11 $\text{cm}^{-1}/\text{bar}$ at room temperature for C_2H_2 and C_2H_6 , extrapolated to other temperatures by assuming that the collision widths vary as temperature to the -0.75 power. A value of 10 cm^{-1} is used for the maximum extent of a spectral line. Opacity due to $\text{H}_2\text{-H}_2$ and $\text{H}_2\text{-He}$ collision induced transitions are computed following the formalism of Borysow *et al.* (1988).

The suite of models listed in Table II is designed to examine the range of possible altitude profiles for C_2H_2 and C_2H_6 . For clarity, we show in Figs. 3 and 4 only those models using the best-fit temperature profile, but Table II lists models for warmer and cooler temperatures. Examination of Figs. 3 and 4 reveals that the models differ in the relative amounts of C_2H_2 and C_2H_6 at high (>10 mbar) and low (<1 mbar) pressure. In constructing these distributions we have been guided by previous analyses. The C_2H_2 distributions lie within the range determined from the UV studies of Wagener *et al.* (1985) and B etremieux and Yelle (2000) in the 10–20 mbar region. All of the models contain far less C_2H_2 at low pressure than predicted by the photochemical models (Gladstone *et al.* 1996). The C_2H_6 mole fractions at 1–20 mbar agree with the results of previous IR and UV investigations (Table I) and, interestingly, the photochemical models (Gladstone *et al.* 1996).

Figures 5 and 6 show the spectral fits obtained with model A. Our other models produce fits of comparable quality. Model A is most consistent with prior analyses and, in fact, resembles the distributions determined by Fouchet *et al.* (2000), despite the different assumptions about temperature and the different latitude coverage of the observations. Figures 5 and 6 also show synthetic spectra based on Gladstone *et al.*'s (1996) photochemical models. Clearly, Gladstone *et al.*'s (1996) C_2H_2 distribution is at odds with the observations, but the C_2H_6 distribution is in excellent agreement.

We estimate the atmospheric levels probed by the emissions with calculations of the contribution function at the centers of the features, convolved to the spectral resolution of the data. For C_2H_2 , this function displays interesting structure (Fig. 7). There are significant contributions at both the 10 mbar and the 0.01 mbar levels, resulting from the pressure-broadened wings of the lines and the saturated line center, respectively. We find that if we adopt the values for the C_2H_2 mole fractions of $8.3\text{--}10 \times 10^{-7}$ and $2\text{--}4 \times 10^{-8}$ at 0.3 and 10 mbar (Fouchet *et al.* 2000, Wagener *et al.* 1985, and B etremieux and Yelle 2000), we determine a C_2H_2 mixing ratio of $1.1\text{--}4.3 \times 10^{-6}$ at 0.01 mbar. The C_2H_6 contribution function (Fig. 7) is simpler in form than the C_2H_2 function because the lines are optically thin. There is a single peak at 5 mbar where the C_2H_6 mole fraction is $2.8\text{--}6.5 \times 10^{-6}$. Because the contribution function for this band peaks at relatively deep pressures, we have no strong constraints on the C_2H_6 abundance at lower pressures, where the problem between observations and photochemical models of C_2H_2 was found.

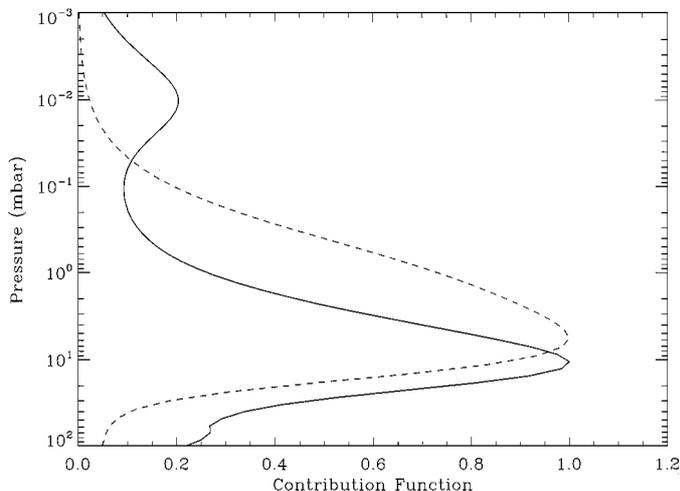


FIG. 7. The contribution functions for the emissions from C_2H_2 (solid line) and C_2H_6 (dashed line) shown in Figs. 5 and 6.

3.5. Aerosol Distribution

The final ingredient needed to calculate the thermal structure is the distribution and optical properties of aerosols. We utilize the results of Moreno (1996), determined from their analysis of multifilter HST images of Jupiter, as given in their Table III.

4. ENERGY BALANCE CALCULATIONS

Our calculations of Jupiter's thermal structure resemble those used to study Titan by Yelle (1991). Radiative processes include heating through absorption of solar energy in CH_4 bands from 1.0 to 4.0 μm , heating by absorption of visible sunlight by aerosols, cooling by emissions from mid-IR vibrational bands of CH_4 , C_2H_2 , and C_2H_6 , and collisionally induced emission from $\text{H}_2\text{-H}_2$ and $\text{H}_2\text{-He}$. In addition to these radiative processes, the thermal structure calculations include thermal conduction. Calculations are carried out for low-latitude regions by averaging the solar heating rates over a latitude circle at 6°N . We do not include seasonal effects, but these are small at equatorial latitudes. The calculations extend from 10^{-3} to 100 mbar with a resolution of 10 levels per decade of pressure.

4.1. Solar Heating Bands

The heating rate resulting from absorption of sunlight in the CH_4 bands at near IR wavelengths is most easily calculated by using k -coefficients to represent the opacity of the atmosphere. There are a number of sources for these coefficients. Recently, Irwin *et al.* (1996) published k -coefficients for the CH_4 bands that are based on Strong *et al.*'s (1993) laboratory measurements of CH_4 transmission spectra. Also, Baines *et al.* (1993) has derived k -coefficients, over a more limited spectral region, from the laboratory measurements of Giver *et al.* (1990). Finally, k -coefficients may be calculated directly from a line list, such as

HITRAN Rothman *et al.* 1992) or GEISA (Husson *et al.* 1991). We use all three of these sources in our calculations.

The good agreement between laboratory measurements and synthetic spectra for the $3.3\ \mu\text{m}$ band of CH_4 suggests that available line lists are fairly complete (Strong *et al.* 1993). This is fortunate because the lowest pressure in the Strong *et al.* (1993) study is 0.268 mbar and the lowest column abundance $1.723\ \text{cm-amagat}$, while the peak heating in the jovian atmosphere from the $3.3\text{-}\mu\text{m}$ band occurs at 0.001 mbar and a column abundance of $0.06\ \text{cm-amagat}$, well outside the range of the Strong *et al.* (1993) measurements. The Giver *et al.* (1990) measurements did not encompass the $3.3\text{-}\mu\text{m}$ band. We therefore use the k -coefficients derived from line-by-line calculations based on the HITRAN line list for the $3.3\text{-}\mu\text{m}$ band.

To evaluate the LBL k -coefficients, we adopt a collisional broadening parameter of $0.07\ \text{cm}^{-1}\ \text{bar}^{-1}$ at 296 K, assumed to vary as temperature to the power of -0.75 . The maximum line extent is set at $10\ \text{cm}^{-1}$. The spectrum is divided into $10\ \text{cm}^{-1}$ intervals and sampled at the Doppler width of a CH_4 line at the coldest temperature in the atmosphere. Ten k -coefficients are used for each interval, distributed as five Gauss–Legendre quadrature points, each between cumulative probabilities of 0 and 0.95 and between 0.95 and 1.00. The dense sampling at large absorption coefficients allows us to accurately represent opacities in the cores of strong lines. The k -coefficients are tabulated at 61 pressures, spaced logarithmically between 10^{-3} mbar and 10^3 mbar and at 11 temperatures spaced evenly in $1/T$ from 50 to 500 K. Interpolation yields the k -coefficients at arbitrary pressures and temperatures.

Available line lists are not adequate for the other near IR bands of CH_4 and the k -coefficients based on laboratory measurements must be used. The Irwin *et al.* (1996) coefficients have the advantage of wider wavelength coverage and of being based on laboratory conditions in which broadening was primarily by H_2 . However, the Strong *et al.* (1993) and Irwin *et al.* (1996) investigations were directed primarily toward the troposphere and, in general, concern higher pressures and temperatures than those that pertain to the jovian stratosphere. The principle advantage of the Baines *et al.* (1993) coefficients is their use of the Giver *et al.* (1990) laboratory measurements, which extend to lower temperatures than do Strong *et al.*'s (1993). The lowest temperature measured by Giver *et al.* (1990) was 112 K, which is close to the temperature at the jovian tropopause. The lowest temperature measured by Strong *et al.* (1996) was 190 K, significantly warmer than the temperature at the jovian tropopause.

Figures 8 and 9 show transmission spectra and heating rates calculated using the Irwin *et al.* (1996) and Baines *et al.* (1993) coefficients. The transmission found using the Baines *et al.* (1993) coefficients is significantly less than that calculated with the Irwin *et al.* (1996) coefficients and the heating rate is significantly higher in the 0.02–3 mbar region. However, it is not clear that the Baines *et al.* (1993) coefficients are more accurate than those of Irwin *et al.* (1993) for this region of the jovian atmosphere. There are many differences between the two studies

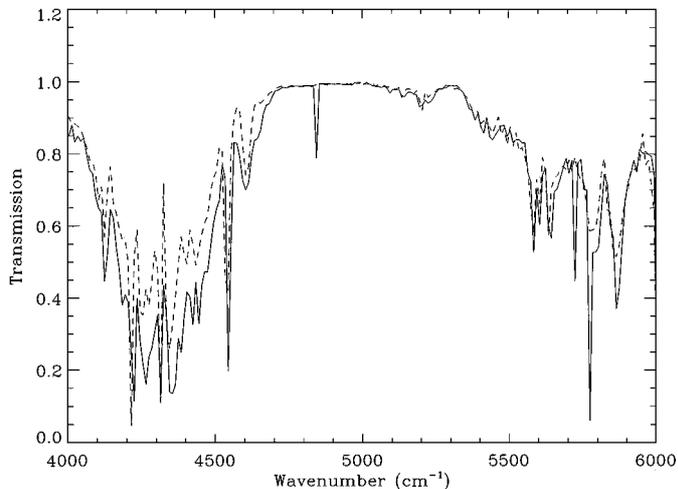


FIG. 8. Vertical transmission to 100 mbar using the Irwin *et al.* (1996) (solid line) and Baines *et al.* (1993) (dashed line) k -coefficients.

in the way that k -coefficients were determined and in the measurements themselves and, in our view, it is difficult to judge between the two. We chose the Baines *et al.* (1993) values because they are based on low-temperature measurements, but it is probably best to view the differences between Baines *et al.* (1993) and Irwin *et al.* (1996) as a guide to the uncertainty in the solar heating rate. The only coefficient available for the 1.1 and $1.3\ \mu\text{m}$ bands are those of Irwin *et al.* (1996), and we use these in our calculations. The choices made here are essentially the same as those of Moreno and Sedano (1997).

Information on the spectral line composition of the visible bands is lacking, thus we use the empirical absorption coefficients derived by Karkoschka (1994). The drawback to this approach is that bands are described by a single absorption

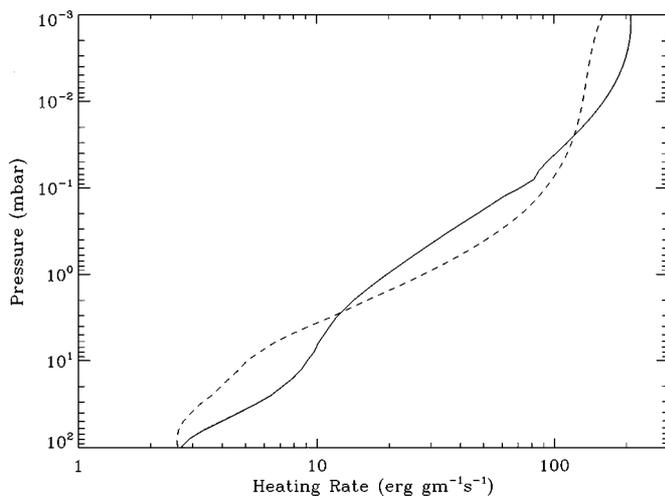


FIG. 9. Heating rates resulting from absorption in the $4000\text{--}5000\ \text{cm}^{-1}$ spectral region, calculated using the Irwin *et al.* (1996) (solid line) and Baines *et al.* (1993) (dashed line) k -coefficients.

coefficient (i.e., Beer's law) that is assumed to be independent of temperature and pressure. This is incorrect, but until our knowledge of the structure of these bands improves, no other approach is possible. The Karkoschka coefficients are determined partly through observations of Jupiter, and our calculated heating rates are therefore consistent with the observed albedo.

Aerosol absorption is modeled by assuming that the aerosols behave as Mie scatterers. Multiple scattering is included at visible wavelengths because the absorption bands are weak, and scattering by both H₂ and aerosols cannot be neglected. Our radiative transfer calculations employ the DISORT routine developed by Stamnes and collaborators (Stamnes *et al.* 1988). Cross sections for Rayleigh–Raman scattering by H₂ are obtained from Ford and Brown (1970). These calculations encompass the 0.2–1.0 μm region. Absorption of sunlight at wavelengths less than 0.2 μm is not treated explicitly but its effects are considered by specifying a downward conductive heat flux of 0.5 $\text{erg cm}^{-2} \text{s}^{-1}$ at the upper boundary of the calculations (Young *et al.* 1997).

4.2. Mid- and Far-IR Cooling Bands

Cooling occurs through radiative emissions from the vibrational bands of CH₄, C₂H₂, and C₂H₆ in the mid-IR (700–1400 cm^{-1}) and collisionally induced transitions of H₂–H₂, He in the mid to far IR (100–800 cm^{-1}). Our calculation follows the procedure described by Borysow *et al.* (1988) to model collisionally induced (CIA) transitions.

The opacities in the mid-IR vibrational bands at 1304 cm^{-1} (CH₄), 729 cm^{-1} (C₂H₂), and 822 cm^{-1} (C₂H₆) are represented with spectral mapping transformations (West *et al.* 1990). The absorption coefficients used to generate the spectral mapping transformations are calculated with the same techniques and line broadening parameters as for the near IR bands, but a maximum line extent of 1 cm^{-1} is used to reduce run time.

We find that the cooling rates in the stratosphere are sensitive to the spectrum of radiation at the lower boundary of the calculation. The usual procedure of specifying that the radiation field at the lower boundary equals the Planck function at the temperature of the lower boundary produces unacceptably large errors. This is due to the fact that the atmosphere below our lower boundary has significant temperature variations over distances of the order of a photon mean free path. Thus, we calculate the intensity field at the lower boundary by an integration through the atmosphere from 100 mbar to a depth of 10 bar.

4.3. Thermal Conduction

Thermal conduction becomes important at pressures less than 0.001 mbar. Thermal conduction parameters are obtained from Hanley *et al.* (1970). The observed temperature profile requires a downward conduction flux of 0.5 $\text{erg cm}^{-2} \text{s}^{-1}$ at the upper boundary of our calculations. This exceeds the solar UV energy deposited in the thermosphere and may be due to viscous dissipation of gravity waves in the upper atmosphere (Young *et al.* 1997).

5. RESULTS

Heating rate calculations are shown in Fig. 10. These rates apply to all models, because they depend only on the CH₄ and aerosol distributions, which are identical in all models. We subdivide the net heating rate into contributions from each of the strong CH₄ bands as well as the visible region, where heating is due partly to aerosols. Because the solar flux increases with decreasing wavelength while the band strengths decrease, the CH₄ bands form a progression with the longer wavelength bands dominating at low pressures and shorter wavelength bands dominating at deeper levels. The 3.3- μm band is the primary heat source at pressures less than 0.004 mbar, the 2.3- μm band between 0.004 and 0.5 mbar, the 1.7- μm band between 0.5 and 2 mbar, and the 1.3- μm and 1.1- μm bands at pressures from 2 to 25 mbar. The total heating rate is a smooth function of pressure from 0.004 to 10 mbar, varying approximately as $p^{-0.3}$.

At no level in the stratosphere is the solar heating dominated by optically thin absorption bands, which would produce a heating rate proportional to pressure. Conrath *et al.* (1990) argued that the lack of latitudinal gradients in the stratospheric temperature is due to the fact that the weak CH₄ bands are optically thin, implying a latitude-independent heating rate. The optically thin behavior of the heating rates in Conrath *et al.* (1990) is a consequence of their approximate radiative transfer treatment. The small latitudinal temperature gradients on Jupiter requires a different explanation.

Absorption in the visible region contributes less than 20% of the total heating at all levels and consequently is relatively unimportant to the stratospheric thermal structure. It appears that most of this heating is due to absorption by CH₄. Although, in a scattering atmosphere, we cannot separate the effects of aerosol and CH₄ absorption, we find that doubling the density of aerosols increases the heating rate by less than 1%. Our finding that aerosol heating is relatively unimportant agrees with the calculations of Moreno and Sedano (1997) for equatorial latitudes, but is at odds with earlier studies of the jovian stratosphere (Cess and Chen 1975, Appleby and Hogan 1984). The difference is likely due to improvements in CH₄ band parameters and radiative transfer techniques. A recent study of the Uranian stratosphere also found aerosol heating to be minor (Marley and McKay 1999), though it dominates in the stratosphere of Titan (McKay *et al.* 1989).

The cooling rates for all composition models share some general features that we illustrate by discussing results for model A, shown in Fig. 11. CH₄ dominates cooling at pressures less than ~ 0.004 mbar, C₂H₂ is important and sometimes dominant in the 0.01 mbar region, and C₂H₆ dominates in a broad region beginning at several hundredths of a millibar and extending to nearly 10 mbar. The relative importance of the hydrocarbon molecules is a consequence of their abundances and opacity structures. Even though it is by far the most abundant, CH₄ plays a relatively minor role in the cooling rates because its spectral lines are strongly saturated, as a result of its large abundance, and this greatly reduces the cooling efficiency. CH₄ is important at low

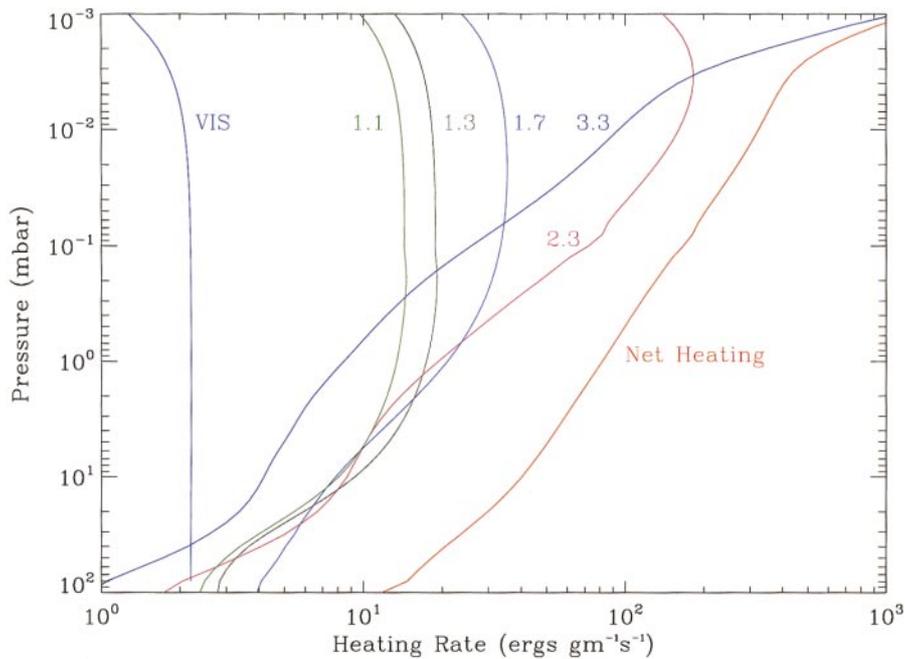


FIG. 10. Heating rates in the near IR bands of CH₄ and for absorption of visible radiation by CH₄ and aerosols.

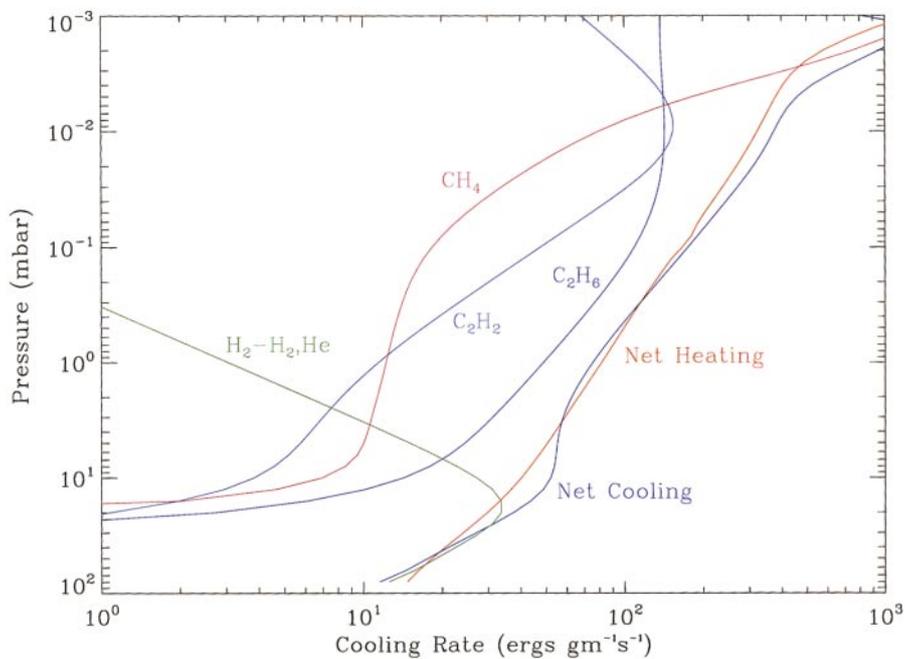


FIG. 11. The contributions to the net cooling rate for model A. C₂H₆ is the dominant coolant throughout most of the stratosphere. The net heating and cooling rates balance to good accuracy.

pressures where some photons from the ν_4 band are able to escape the atmosphere, but below 0.004 mbar the large overlying abundance of CH_4 effectively traps outgoing photons. CH_4 also suffers because the overlap between the ν_4 band and the Planck function is weak compared with the C_2H_2 and C_2H_6 bands. Throughout most of the stratosphere, C_2H_2 and C_2H_6 are the most important coolants. The C_2H_2 cooling rate decreases with increasing pressure faster than that for C_2H_6 partly because its mole fraction decreases more quickly with pressure and partly because of differences in opacity structure. Although the C_2H_2 and C_2H_6 bands are similar in strength and fairly close in wavelength, the C_2H_6 band is diffuse with many lines over a relatively large wavelength region; the C_2H_2 band is compact with most of the band strength concentrated in a small number of lines. As a consequence, the C_2H_2 lines tend to be optically thick and are unable to cool effectively. In contrast, C_2H_6 is optically thin throughout the stratosphere.

This new result, that cooling by C_2H_6 dominates the energy balance in the stratosphere, becomes apparent when a realistic C_2H_6 altitude distribution is considered, rather than the constant mole fraction distributions assumed in earlier studies.

Cooling because of the CIA opacity of $\text{H}_2\text{-H}_2$ and $\text{H}_2\text{-He}$ dominates at pressures greater than ~ 10 mbar. Careful examination of Fig. 11 shows that the net cooling rate is actually smaller than the CIA cooling rate. This is because the net cooling rate has been defined as the sum of the contributions of CIA, CH_4 , C_2H_2 , and C_2H_6 . All three hydrocarbons actually heat the atmosphere near 100 mbar, because they are absorbing warmer radiation from deeper in the troposphere. This effect only appears when the radiation field at the lower boundary is accurately calculated. Once again, C_2H_6 has the largest effect and contributes a heating of $2 \text{ erg gm}^{-1} \text{ s}^{-1}$ at 100 mbar, roughly 10% of the net heating rate. Obviously, adding together the hydrocarbon and CIA terms muddles the definition of net cooling rate, but it seems to be the only sensible division.

The comparison of net heating and net cooling rates in Fig. 11 shows that the stratosphere is close to radiative equilibrium. The heating rate exceeds the cooling rate by about $5 \text{ erg gm}^{-1} \text{ s}^{-1}$ at 100 mbar and falls below the cooling rate by about $10 \text{ erg gm}^{-1} \text{ s}^{-1}$ at 10 mbar. Larger differences appear at pressures less than 0.003 mbar but at these levels thermal conduction and dissipation of mechanical waves are thought to be important (Young *et al.* 1997), and radiative equilibrium is not expected. The disagreement between the heating and cooling rate is larger if the Irwin *et al.* (1996) coefficients are used to calculate the heating rate in the 1.7 and 2.3 μm bands; however, it would be incorrect to suggest that the net radiative imbalance in the models implies an imbalance in the atmosphere. In our view, the difference between heating rates based on Baines *et al.* (1993) and Irwin *et al.* (1996) implies that we are unable to calculate the heating rates to better than a few $\text{ergs gm}^{-1} \text{ s}^{-1}$. If one adopts this view, then it follows that the stratosphere is in radiative equilibrium to within a few $\text{erg gm}^{-1} \text{ s}^{-1}$, and there is no evidence to suggest significant departures.

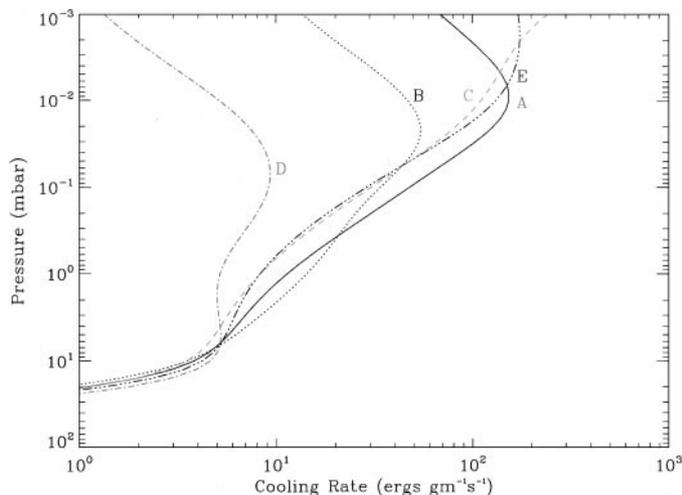


FIG. 12. Calculated cooling rates for different models of the C_2H_2 distribution described in Table II. For clarity we show only a subset of the models—the other models are very similar.

The uncertainties in the C_2H_2 and C_2H_6 altitude distributions imply relatively minor uncertainties in the radiative cooling rate. Figures 12 and 13 show calculations of the net cooling rates, based on the composition models in Table II, that illustrate the uncertainty because of C_2H_2 and C_2H_6 , respectively. It is interesting that the range of possible C_2H_2 distributions produce cooling rates that are generally smaller than the nominal rate (model A) while the range of possible C_2H_6 distributions produce cooling rates that are generally larger than the nominal rate. Obviously, we could improve the match between heating and cooling rates by tweaking the C_2H_2 and C_2H_6 altitude distributions, but such an exercise would have only cosmetic value.

The disagreement between heating and cooling rates near 30 mbar is independent of composition, because $\text{H}_2\text{-H}_2$ CIA

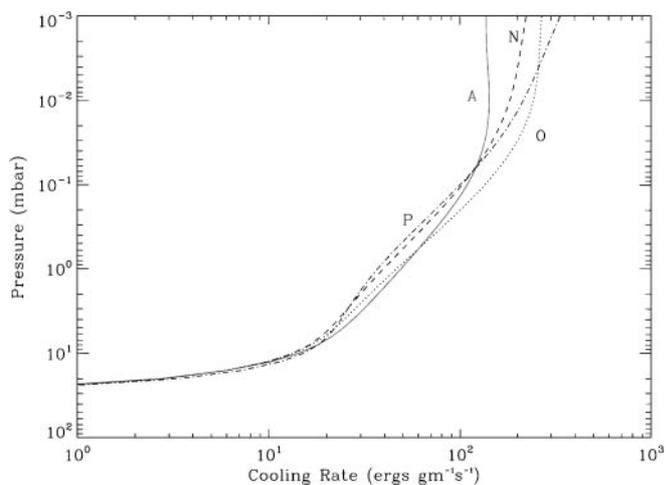


FIG. 13. Calculated cooling rates for different models of the C_2H_6 distribution described in Table II. For clarity we show only a subset of the models—the other models are very similar.

cooling dominates in this region. Unfortunately, it is difficult to draw conclusions from this disagreement because the net heating rates are uncertain in this region by an amount comparable to the calculated imbalance. Improvements in our knowledge of the CH_4 absorption coefficients are needed before this feature can be interpreted with confidence.

The radiative balance in the stratosphere can also be studied through comparison of radiative equilibrium temperatures with the measured temperatures. Figure 1 shows a radiative equilibrium temperature profile based on composition model A. This profile is calculated by adjusting the temperature in an iterative fashion until the net heating rates are 10^{-3} times smaller than either the heating or cooling rate alone at all levels (Yelle 1991). The calculated profile fits the ASI measurements nearly as well as the empirical models derived in Section 3.1. This reinforces the conclusion that, to within the accuracy of our calculations, the jovian equatorial stratosphere is in radiative balance.

6. DISCUSSION

The atmospheres in this solar system with well-developed stratospheres include the Earth, Jupiter, the other giant planets, and Titan. Neither Mars nor Venus has a warm, stable atmospheric region above the tropopause that resembles the terrestrial stratosphere. Titan's aerosol-dominated stratosphere differs significantly from the others, where gas absorption and emission dominates. Thus, we have the interesting situation in that the stratosphere of the Earth has its closest analogue not in the atmospheres of its terrestrial cousins but in the atmospheres of the giant planets.

The terrestrial stratosphere is cooled primarily by CO_2 emissions and heated primarily through absorption of sunlight by O_3 , a product of O_2 photochemistry. Thus, Earth and Jupiter share the circumstance that photochemically produced species are important in the thermal structure. Latitudinal variations in O_3 play a large role in the dynamics of the terrestrial stratosphere, and it seems likely that latitudinal variations in C_2H_6 play a large role in the dynamics of the jovian stratosphere. A particularly intriguing possibility is that the intense jovian aurora, by synthesizing C_2H_6 and other thermally active hydrocarbons, could have a significant effect on the global dynamics of the stratosphere.

It follows that knowledge of the C_2H_2 and C_2H_6 chemistry is required to understand the jovian stratosphere. However, the C_2H_2 abundance determined through analysis of the Irshell observations is significantly smaller than predicted by the photochemical models of Gladstone *et al.* (1996). Similar conclusions regarding jovian C_2H_2 have been reached by Bézard *et al.* (1997b) and Fouchet *et al.* (2000). Our baseline model has a C_2H_2 mole fraction nearly 20 times smaller than that of Gladstone *et al.* (1996) at 0.01 mbar. Previous attempts to understand the low C_2H_2 abundance have postulated chemical destruc-

tion of C_2H_2 by H atoms (Allen *et al.* 1992, Gladstone *et al.* 1996, Romani 1996). However, even the model by Romani (1996), which incorporates the most aggressive C_2H_2 destruction scheme, only decreases the C_2H_2 mole fraction by a factor of 4 from that calculated by Gladstone *et al.* (1996). An even more efficient C_2H_2 destruction mechanism is needed, or perhaps C_2H_2 is not produced as efficiently as currently calculated in the photochemical models.

How similar are the energetics and dynamics of the terrestrial and jovian stratospheres? This study shows that the equatorial regions on Jupiter are close to radiative equilibrium. The net radiative imbalance estimated here of $<10 \text{ erg gm}^{-1} \text{ s}^{-1}$ corresponds to a heating rate of $<10^{-3} \text{ K}$ per jovian day. Heating and cooling rate in the terrestrial stratosphere are well balanced at all but polar latitudes, with net rates of $<1 \text{ K}$ per earth day (Mlynczak *et al.* 1999a, 1999b). Though most clearly demonstrated in the Mlynczak *et al.* (1999a, 1999b) studies, this observational result has been known for some time. The accuracy of radiative balance was a mystery until Dunkerton (1978) explained it as a consequence of the Charney–Drazin theorem, which states that the energy transported by waves is precisely balanced by the energy carried away by the circulation established by the waves (Charney and Drazin 1961, Boyd 1976, Andrews and McIntyre 1976). This theorem applies as long as the effects of transience, nonlinearities, and dissipation are small, and this is found to be the case in the terrestrial stratosphere. This result, though profound, is primarily observational, because it is notoriously difficult to calculate the nonlinear and transient effects. Usually their presence can only be deduced from observations. Thus, we cannot say with confidence whether the radiative dominance of the terrestrial stratosphere is a general result, linked to the Charney–Drazin theorem, or peculiar to the Earth.

Observations of several planetary stratosphere are required to determine if radiative dominance is a general rule. The Friedson *et al.* (1999) study finds that aerosols are transported primarily by eddies associated with a wave-driven circulation, implying that the jovian stratosphere differs fundamentally from that of the Earth. However, it would be premature to adopt this as a firm conclusion until latitudinal variations in C_2H_2 and C_2H_6 are included in dynamical models of the jovian stratosphere.

7. CONCLUSIONS

Our conclusions can be summarized as follows:

1. C_2H_6 is the primary coolant in the equatorial jovian stratosphere in the 0.05–10 mbar pressure region. This result differs from previous analyses and is a consequence of considering realistic density distributions for C_2H_6 . It is interesting that the energetics of the jovian stratosphere are dominated by a photochemically produced molecule.
2. Latitudinal variations in the abundance of C_2H_2 and C_2H_6 between 0.004 and 10 mbar must be taken into account in

dynamical models. As the primary thermally active species, they could play a large role in forcing stratospheric dynamics.

3. The stratosphere is close to radiative balance in the 0.03–100 mbar region. The net radiative heating rate is of order or smaller than the uncertainties in heating rate calculations.

4. Significant uncertainties still exist in the absorption coefficients for the near IR bands of CH₄. Better data are needed to improve our ability to accurately calculate solar heating rates.

5. The mole fraction of C₂H₂ in the 0.004–0.02 mbar region and C₂H₆ in the 0.4–10 mbar region are $1.1\text{--}4.3 \times 10^{-6}$ and $2.8\text{--}6.5 \times 10^{-6}$, respectively. This C₂H₂ abundance is much less than predicted by photochemical models; the C₂H₆ abundance is in good agreement.

These conclusions imply that the photochemistry, radiative balance, and dynamics of the jovian stratosphere are strongly coupled and must be treated in a self-consistent manner.

APPENDIX: TEMPERATURE-INDEPENDENT FORMULATION OF THE DIFFUSION EQUATION

The usual form of the diffusion equation is

$$\Phi_i = -D \left(\frac{dN_i}{dz} + \frac{N_i}{H_i} + \frac{N_i}{T} \frac{dT}{dz} \right) - K \left(\frac{dN_i}{dz} + \frac{N_i}{H_a} + \frac{N_i}{T} \frac{dT}{dz} \right). \quad (\text{A1})$$

Changing variables from number density N_i to mole fraction f_i and from altitude z to Z . Where $Z = -\log(p/p_0)$ and p_0 is a reference pressure, we have

$$-\frac{\Phi_i H_a}{(D+K)N_a} = \frac{df_i}{dZ} + \frac{(m_i/m_a - 1)D}{D+K} f_i, \quad (\text{A2})$$

where N_a is the total number density and H_a the pressure scale height. We assume that the molecular diffusion coefficient varies as $D = g(T)p_0/p$ and the eddy diffusion coefficient as $K = g(T)(p_0/p)^\gamma$ so that the ratio D/K depends only on pressure: $D/K = r(p_0/p)^{1-\gamma}$ where r is a constant and p_0 a reference pressure. The function $g(T)$ is arbitrary, but we assume that K and D have this same temperature dependence. We now restrict our attention to cases where the flux is zero. In this case, Eq. (A2) has a simple analytic solution with no explicit temperature dependence,

$$f(p) = f(\infty)(1 + r(p_0/p)^{1-\gamma})^{\frac{1-m_i/m_a}{1-\gamma}}. \quad (\text{A3})$$

The assumptions leading to this result are not restrictive. Because the eddy diffusion coefficient is often treated as a parameter to be determined by fitting observations, nothing is lost by assuming a specific temperature dependence. There is usually sufficient freedom with the pressure dependence to fit any observations, but if the two parameter model for K adopted here is found to be inadequate, more complex profiles can be constructed by joining solutions for different regions in a piecewise continuous fashion. The lack of an explicit temperature dependence means that A3 can be used in nonisothermal atmospheres and makes this approach particularly useful for thermal structure calculations.

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